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PATENT SPECIFICATION

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(54) BASICALLY SUBSTITUTED BENZYL-PHTHALAZINONE DERIVATIVES, ACID ADDITION SALTS THEREOF AND PROCESS FOR THE PRODUCTION THEREOF

We, ASTA-WERKE AKTIEN-(71)GESELLSCHAFT CHEMISCHE FABRIK, company organised under the laws of Germany of, 4812 Brackwede, West-5 falen, Germany, do hereby declare the invention, for which we pray that a patent may be granted to us, and the method by which it is to be performed, to be particularly described in and by the following statement: —

The present invention is related to new basically substituted benzyl-phthalazinone derivatives having a high antihistamine effectiveness, the physiologically acceptable acid addition salts thereof, and a process for the

production thereof.

The new benzyl-phthalazinone derivatives according to the present invention are characterized by a cyclic basic residue which is connected to the nitrogen atom in the position 2 of the phthalazinone nucleus by a carbon atom of this cyclic basic residue directly or by way of an alkylene chain. Basically substituted phthalazinones are known already, for instance from German patent specification No. 1,046,625. These phthalazinones are compounds having a basic residue substituted on an alkylene chain, this basic residue being derived from a tertiary amine having two alkyl groups or an alkylene group (thus 30 forming a cyclic residue). However, the cyclic basic residue is connected to the nitrogen atom in the position 2 of the phthalazinone nucleus by the nitrogen atom of the amine by way of the alkylene chain.

Accordingly the present invention provides a basically substituted benzyl-phthalazinone derivative of formula I

[Price 25p]

$$(R_2)_n$$
 $(R_1)_m$
 $(R_2)_n$
 $(X)_p$
 $(X)_p$

wherein R₁ and R₂, which may be identical or different from each other, represent a hydrogen or halogen atom, an alkyl radical containing from 1 to 4 carbon atoms, an alkoxy radical containing from 1 to 4 carbon atoms, or a hydroxy, trifluoromethyl, nitro or substituted or unsubstituted amino group, X is an alkylene group having the formula

$$-CH_2$$
— or $-CH$ —,

m and n, which may be identical or different from each other, represent 1, 2 or 3, p is 0 or 1, and the group

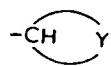
N-C₁₋₄ - alkyl - substituted pyrrolidinyl, an $N-C_{1-4}$ alkyl - subDERWENT PUBLICATION

stituted piperidyl, an N-C, - alkylsubstituted perhydroazepinyl, quinuclidinyl,
tropanyl or scopyl group, the tropanyl
or scopyl group being connected to
the 2-nitrogen atom of the phthalazinone directly by way of a ring carbon
atom of this tropanyl or scopyl group, and
the other groups being connected to the 2nitrogen atom of the phthalazinone directly or
by way of an alkylene group of the formula

and their physiologically acceptable acid addition salts.

those compounds of formula I and their physiologically acceptable acid addition salts are preferred wherein R₁ and R₂ represent hydrogen, halogen, hydroxy, an alkyl group containing from 1 to 4 carbon atoms, an alkoxy group containing from 1 to 4 carbon atoms or trifluoromethyl, and m and n are 1 or 2. Particularly preferred are those compounds of this preferred group wherein R₁ represents such an atom or group as indicated above and R₂ is a hydrogen atom.

The most preferred group of compounds of formula I and their physiologically acceptable acid addition salts comprises those compounds wherein R₁ is a hydrogen, fluorine, chlorine or bromine atom or a methoxy, ethoxy, methyl, hydroxy or trifluoromethyl group, R₂ is a hydrogen atom, m is 1 or 2, p is 0, and the group



is the N-methyl-perhydroazepinyl, the tropanyl or the quinuclidinyl group, in particular the N-methyl-perhydroazepinyl-(4), the tropanyl-(3) or the quinuclidinyl (3) group. Thus, the fused benzene ring of these benzyl-phthala-zinone derivatives is unsubstituted and the perhydroazepinyl, tropanyl or quinuclidinyl residue is connected directly to the 2-nitrogen

atom of the phthalazinone nucleus.

The process for producing the new, basically substituted benzyl-phthalazinone derivatives of formula I and the physiologically acceptable acid addition salts thereof is characterized in that A) a compound of formula II

or a reactive derivative thereof, wherein R₁, R₂, m and n have the same meanings as in formula I, is subjected to reaction with a hydrazine of formula III

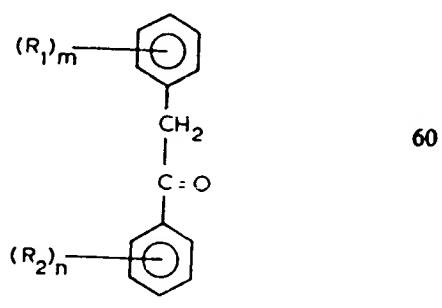
$$H_2N-NH-R_3$$

III

II

wherein R₃ is hydrogen or the group 55

wherein X, p and —CHY have the same meanings as in formula I, or B) a compound of formula IV



IV

wherein R₁, R₂, m and n have the same meanings as in formula I, is subjected to reaction with a compound of formula V

V

wherein R₂ has the same meaning as in formula III and R₄ is an alkyl group containing from 1 to 4 carbon atoms, or

VI

wherein R₂ and n have the same meanings as in formula I and R₃ has the same meaning as in formula III, is subjected to reaction with a compound of formula VII

VII

wherein R₁ and m have the same meanings as in formula I and Z is a halogen atom or a hydroxy or alkoxy group, or D) when

15 is an

N—C_{1→} alkyl-substituted pyrrolidinyl, N—C_{1→} alkyl-substituted piperidyl or N—C_{1→} alkyl-substituted perhydroazepinyl group,

20 a compound of formula I wherein the group

is a pyrrolidinyl group, a piperidyl group or a perhydroazepinyl group is subjected to reaction with an alkylating agent containing from 1 to 4 carbon atoms and subjecting a benzyl-phthalazinone derivative of formula XIII

wherein R₁, R₂, m and n have the same meaning as in formula I and R₃ is hydrogen, resulting from process A), B), or C), to reaction with a compound of formula VIII

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wherein Q represents an atom or group which, upon substitution of the amide group, is split off together with its electron doublet, such as a halogen atom or a sulphonic ester group, and R₃ is the group

X, p and

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having the same meanings as in formula I, and converting the thus obtained benzyl-phthalazinone derivative, if desired, with an appropriate acid into a physiologically acceptable acid addition salt thereof or if desired, converting a resulting salt of a benzyl-phthalazinone derivative into the free base.

A reactive derivative of the carboxylic acid of formula II is in particular an acid halide, ester or anhydride. Other reactive derivatives of the compounds of formula II which may be used are the unsaturated or saturated phthalides or phthalimidines of the formula X

and XI

$$(R_2)_{n}$$
 $(R_1)_{m}$
 $(R_1)_{m}$

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In the above formulas X and XI, R₁, R₂, m and n have the same meanings as in formula I and A is an oxygen atom or imino group and R₃ is halogen, NH₂, ArNH (wherein Ar represent an aryl group), OH or an alkoxy group. Other compounds of this type are those of formula XII

XII

wherein R₁, R₂, m and n have the same meanings as in formula I. These compounds produce derivatives of the compound of formula II when subjected to reaction with a compound of formula III.

The above procedures A, B and C are carried out in the absence or presence of usual solvents and auxiliary agents at a temperature elevated up to about 180° C. and in a pH range varying from acidic to alkaline.

Useful solvents are, for instance, water, alcohols, dimethyl-formamide, dioxane, pyridine, triethylamine and hydrocarbons. Useful auxiliary agents are bases, acids and condensation agents usual for such reactions.

The procedure D is carried out with usual alkylating agents such as formaldehyde in the presence of a reducing agent such as formic acid, NaBH, or hydrogen, as well as dimethyl sulfate and K₂CO₃, alkyl halides or diazomethane. The reaction E preferably is carried out with catalytic hydrogen. Useful catalysts are preferably the precious metal and nickel catalysts.

When carrying out the reaction with the alkylating agents of formula VIII, the known cyclammonium rearrangements [See Henecka, Hörlein und Risse "Zur Kenntnis dercyclo-ammonium-Umlagerung" Ang. Chem. 72, 960(1960] may take place with a change in the ring size.

The compounds of formula I and their acid addition salts are to a great extent optically active by way of the carbon atom of the cyclic basic group which is connected to the amide nitrogen atom of the phthalazinone nucleus directly or by way of an alkylene group. The racemates may be split up into the optically active antipodes in a manner known per se.

The compounds according to the present invention are histaminolytically active. They are characterized by an extremely high activity upon parenteral and above all oral application. They furthermore produce this high activity over a long period of time. This activity may

be shown in the histamine aerosol test on guinea-pigs or in the lesion test in humans, the lesion being caused by histamine or a histamine liberator (Quaddel-Test).

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In guinea-pigs, the histaminolytical activity has been tested in the histamine aerosol test. Guinea-pigs of the Pirbright race weighing 300 to 700 grams each have been tested. The animals inhale an aerosol of an aqueous solution of histamine dihydrochloride in a concentration of 4 mg./ml. The inhalation produces severe dyspnea (severe shortness of breath, lateral positioning) in untreated animals within 2 minutes. In order to determine the histaminolytical activity, the test compounds are applied subcutaneously or orally to groups of 8 to 10 animals. Thereafter, the test animals are treated for varying times with the histamine aerosol. The test animals are considered as protected if they tolerate the inhalation of the aerosol for 10 minutes without showing severe dyspnea (lateral positioning).

For evaluating the test results, the mean effective doses (ED 50 mg./kg.) are determined by means of a probit analysis from the relation between the dose logarithm and the frequency of protection.

Compounds which are similar in chemical structure to the compounds of the present invention and, therefore, have been used for comparative tests, are 4-benzyl-2-(2-dimethylaminoethyl) - 1 - (2H) - phthalazinone ("Ahanon" according to German patent specification No. 1,046,625; compound A in Tables 90 I and II) and β-dimethylaminoethyl-(4-chloro-α-methyl-benzhydryl)-ether, known as a highly active histaminolytic (generic name: chlorophenoxamine; H. Arnold et al., Arzneim-Forsch. 4, 189 (1954); N. Brock et al., Arzneim-Forsch. 4, 262 (1954); compound B in Tables I and II).

The difference between the products according to the present invention and the comparative products A and B is particularly 100 obvious when administering the test compounds to the test animals orally and treating the test animals with the histamine aerosol 8 hours later. Upon application of 0.0215 mg./kg. of 4 - (p - fluorobenzyl) - 2 - [N - methylperhydroazepinyl - (4)] - 1 (2H) - phthalazinone or 0.215 mg./kg. of 4-(p-chlorobenzyl) - 2 - [N - methyl - perhydroazepinyl-(4)] - 1 - (2H) - phthalazinone or 4 - (pchlorobenzyl) - 2 - [quinuclidinyl - (3)] - 1-(2H)-phthalazinone, not one of the 8 to 10 animals of each group showed dyspnea with lateral positioning after treatment with the histamine aerosol. In striking contrast thereto, upon application of 10 to 100 times the dose 115 of both comparative compounds (2.15 mg./kg.) 9 of 10 animals with compound A and 10 of 10 animals with compound B still showed very severe dyspnea with lateral positioning.

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TABLE I

Histaminolytical activity in the histamine aerosol test on guinea-pigs; subcutaneous administration of test compounds 1 hour before treatment with the aerosol

Example No.	ED 50 [mg./kg.]	relative activity (activity of A=1.00)
3	0.0062	17.7
6	0.011	10.0
7	0.0071	15.5
9	0.045	2.44
10	0.031	3.55
11	0.035	3.14
12	0.022	5.00
19	0.016	6.88
24	0.027	4.07
28	0.059	1.86
30	0.026	4.23
33	0.016	6.88
34	0.119	5.79
Α	0.11	1.00
В	0.11	1.00

TABLE II

Histaminolytical activity in the histamine aerosol test on guinea-pigs; oral administration of test compounds 2 and 8 hours before treatment with the aerosol.

Example No.	ED 50 [mg./kg.]		relative activity (activity of A=1.00)	
	2 hours	8 hours	· 2 hours	8 hours
9	0.16	0.49	19.4	13.1
10	0.037	0.029	83.8	221
19	0.010	0.011	310	582
24	0.087	0.052	35.6	123
28	0.20	0.28	15.5	22.9
30	0.038	0.35	81.6	18.3
A	3.1	6.4	1.00	1.00
В	0.52	6.2	5.96	1.03

The histaminolytical activity of the compounds according to the present invention is substantially higher than those of the comparative test compounds A and B. Upon subcutaneous administration, the relative activity is about 17.7 times larger (Example No. 3) than that of the comparative test compounds. The activity is particularly evident upon oral administration (Table II). The activity is 16 to 310 times higher in a 2 hours test in comparison to the activity of test compound A and is 13 to 582 times higher in the 8 hours test. The 8 hours test clearly demonstrates 15 the very high oral activity of the compounds according to the present process which activity is produced over a prolonged period of time.

The compounds according to the present invention are used as active ingredients in 20 pharmaceutical preparations and may be administered in usual embodiments such as tablets, dragees, capsules, suppositories, drops, ointments, creams and injection solutions. They are in particular used for the treatment 25 of the various forms of allergies. Thus, they have been used successfully in humans in the treatment of bronchial asthma, for the treatment of disorders of the skin and mucous membranes such as urticaria, Quincke's edema, pruritus, eczemas, hay fever and rhinitis vasomotorica. In general, they are administered in such treatments in a dosage of 0.4 to 4 mg. per day. The symptoms of the above allergic diseases may be effectively reduced upon a single dose for up to 24 hours. The effectiveness of the compounds of the present invention in humans, which is produced very rapidly

and over a prolonged period of time in comparison to other antihistamines, may be particularly well shown in the reduction of the size of an artificially produced lesion by means of a histamine liberator according to L. Kerp, H. Kasimir, P. N. Tie, Med. Welt 17 NF, 2794 (1966). The compounds according to the present invention may be used as such or in combination with other active ingredients as is usual in antihistaminic preparations. In this respect their minimal dose is most advantageous.

The present invention is further illustrated by the following examples. The constitution of the final products has been verified by elementary analysis and infrared and NMR spectra.

Example 1. 4-Benzyl-2-[N-methylpyrrolidinyl-(3)methyl]-1-(2H)-phthalazinone.

10.3 g. of α-phenylacetophenone-o-carboxylic acid and 6.1 g. of hydrazine sulfate are dissolved in a solution of 3.6 g. of NaOH 60 in 100 cc. of water. The solution is heated to boiling for 2 hours. The precipitate is filtered off with suction, washed with water and dried. The thus obtained 9.2 g. of 4benzyl-1-(2H)-phthalazinone are added to a solution of 1.4 g. of metallic potassium in 250 cc. of anhydrous alcohol. The resulting mixture is heated to boiling for 30 minutes. The alcohol is distilled off. 10.6 g. of the potassium salt are obtained.

12.4 g. of the tosyl ester of 3-hydroxymethyl-N-methylpyrrolidine and 10.6 g. of

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5	the potassium salt of 4-benzyl-1 (2H)-phthal- azinone in 100 cc. of dimethylformamide are heated for one hour at 100° C. The solvent is separated in a rotary evaporator and the residue is triturated with water. The in- soluble matter is dissolved in ether and the
	ethereal solution is extracted with dilute hydro-
0	chloric acid. The acidic extracts are rendered alkaline by the addition of an aqueous potassium hydroxide solution. The separated oily
	product is dissolved in ether and the ethereal solutions are dried over anhydrous Na ₂ SO ₄ .
	Upon evaporation of the ether, 11 g, of the
5	base are obtained. The fumarate crystallizes as the monohydrate. M.p.: 129—132° C.
	The following compounds have been pre- pared in a manner which is similar to the
	procedure of Example 1.
	2. 4 - (p - Chlorobenzyl) - 2 - [N - methyl-
,	pyrrolidinyl - (2) - methyll - 1 (2H) - phthal- azinone hydrochloride.
	M.p.: 206—207° C.
	3. 4 - (p - Chlorobenzyl) - 2 - [N - methyl-
5	piperidyl - (2) - methyl] - 1 (2H) - phthalazi- none sulfate hydrate.

2 M.p.: above 90° C. (with decomposition). 4. 4 - Benzyl - 2 - [N - methylpiperidyl-(3) - methyll - 1 (2H) - phthalazinone hydrochloride hydrate. M.p.: above 77° C. (with decomposition). 5. 4 - (p - Methylbenzyl) - 2 - [N - methylpyrrolidinyl - (2) - methyll - 1 - (2H)-35 phthalazinone hydrochloride hydrate. M.p.: 126—128° C. 6. 4 - (p - Methoxybenzyl) - 2 - [Nmethylpyrrolidinyl - (2) - methyll - 1 - (2H)phthalazinone. M.p.: 111—114° C. 40 7. 4 - (p - Chlorobenzyl) - 2{1 - [Nmethyl - piperidyl - (2)l - ethyl} - 1 - (2H)phthalazinone citrate. M.p.: 103—105° C.

45 Example 8. 4 - Benzyl - 2 - [N - methyl - perhydroazepinyl - (4)l - 1 - (2H) - phthalazinone. A solution of 8 g. of 4-chloro-N-methylperhydroazepine in 20 cc. of toluene is added 50 to a suspension of 13.7 g. of the potassium salt of 4-benzyl-1-(2H)-phthalazinone in 250 cc. of anhydrous toluene dropwise with rapid stirring at 40° C. Heating is applied slowly up to boiling whereafter refluxing is 55 continued for another 5 hours. The solvent is separated in a rotary evaporator and the residue is washed with water. The insoluble oily product is dissolved in ether and the ethereal solution is extracted with dilute 60 hydrochloric acid. The acidic extracts are rendered alkaline by the addition of aqueous potassium hydroxide and the separated oil is again dissolved in ether. The ethercal solutions

are dried over anhydrous Na₂SO₄. Upon evaporation of the solvent, 32 g. of a raw product are obtained. This product is converted into the fumarate which is recrystallized, thus resulting in the fumarate hydrate of the 4 - benzyl - 2 - [N - methyl - perhydroazepinyl-(4)1-1 (2H)-phthalazinone. 70 M.p.: 156—160° C. Example 9. 4-(p-Chlorobenzyl)-1-[N-methyl-perhydroazepinyl-(4)1-1 (2H)-phthalazinone. 30.6 g. of α - (p - chlorophenyl)acetophenone-o-carboxylic acid and 16 g. of hydrazine sulfate are heated with 9.4 g. of NaOH in 250 cc. of water. After washing and drying, 27 g. of 4-(p-chlorobenzyl)-1 (2H)phthalazinone are obtained. 80 20 g. of 2-(2-chloroethyl)-N-methylpyrrolidine hydrochloride are added to a solution of 4.4 g. of NaOH in 20 cc. of water. This solution is heated to 70° C. and added dropwise to a mixture of the above obtained 27 g. of 4-(p-chlorobenzyl)-1 (2H)-phthalazinone and 40 cc. of 50% soda lye heated to 70° C. The mixture is kept at this temperature and heated for another hour. After cooling and diluting with water, the insoluble materials are separated and dissolved in methylene chloride. The solution is extracted with dilute hydrochloric acid and the acidic extracts are rendered alkaline by the addition of aqueous potassium hydroxide. The separated oil is again dissolved in methylene chloride and the solution is dried and evaporated. The crude final product is obtained in a yield above 90% of the theoretical. It is converted into a salt and purified by recrystallization. The hydrochloride of 4-(p-chlorobenzyl)-2-[N-methylperhydroazepinyl-(4)]-1 (2H)-phthalazinone melts at 225-229° C. The following compounds have been prepared in a manner which is similar to the 105 procedure of Examples 8 and 9. 10. 4 - (p - Methylbenzyl) - 2 - [Nmethyl - perhydroazepinyl - (4)l - 1 (2H)phthalazinone sulfate. M.p.: 199—203° C. 110 11. 4 - (p - Methoxybenzyl) - 2 - [N-M.p.: 203—205° C. 12. 4 - (3,4 - Dimethoxybenzyl) - 2 - [N- 115 M.p.: 118-120° C. 13. 4 - (2 - Chlorobenzyl) - 2 - [N-

methyl - perhydroazepinyl - (4)l - 1 - (2H)phthalazinone sulfate. methyl - perhydroazepinyl - (4)| - 1 - (2H)phthalazinone sulfate. methyl - perhydroazepinyl - (4) - 1 (2H)- 120 phthalazinone hydrochloride. F.p.: 198—200° C. 14. 4 - (3 - Chlorobenzyl) - 2 - [N - methylperhydroazepinyl - (4)l - (2H) - phthalazinone. F.p.: 77—78° C. 125 15. 4 - (p - Chlorobenzyl) - 6,7 - dimethoxy - 2 - [N - methyl - perhydro-

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	azeninul - (4)] 1 (24) shthalasinese				
	azepinyl - (4)l - 1 (2H) - phthalazinone	pared in a manner which is similar to the			
	sulfate.	procedure of Examples 23 and 24.			
	F.p.: 286290° C.	25. 4 - Benzyl - 2 - [N - methylpiperidyl-			
_	16. 4 - (2,4 - Dichlorobenzyl) - 2 - [N-	(4)1-1 (2H)-phthalazinone hydrate.			
5	methyl - perhydroazepinyl - (4)l - 1 - (2H)-	M.p.: 106—110° C.			
	phthalazinone fumarate.	26. 4 - (p - Chlorobenzyl) - 2 - [tropanyl-			
	F.p.: 207—211° C.	(3) - 1 (2H) - phthalazinone hydrochloride			
	17. 4 - (p - Dimethylaminobenzyl) - 2-				
	[N - methyl - perhydroazepinyl - (4)] - 1	hydrate.			
10	(2H) phthologicone fumerous	M.p.: 270—274° C.			
10	(2H) - phthalazinone fumarate.	27. 4 - Benzyl - 2 - [quinuclidinyl - (3)]-1			
	F.p.: 177—182° C.	(2H)-phthalazinone fumarate hydrate.			
	18. 4 - (p - Fluorobenzyl) - 2 - [N-	M.p.: 233235° C.			
	methyl - perhydroazepinyl - (4) - 1 (2H)-	28. 4 - (p - Chlorobenzyl) - 2 - [N-			
	phthalazinone sulfate.	methylpyrrolidinyl - (3)l - 1 (2H) - phthal-			
15	F.p.: 211—220° C.	azinone.			
	19. 4 - (p - Bromobenzyl) - 2 - [N-methyl-	M.p.: 117—120° C.			
	perhydroazepinyl - (4)l - 1 (2H) - phthalazi-	20 4 (n. Mothember 1) 2 for in			
	none sulfate.	29. 4 - (p - Methoxybenzyl) - 2 - [quinu-			
		clidinyl - (3)l - 1 (2H) - phthalazinone			
20	F.p.: 215—220° C.	hydrochloride.			
20	20. 4 - (p - Acetylaminobenzyl) - 2 -[N-	M.p.: 236—237° C.			
	methyl - perhydroazepinyl - (4)l - 1 (2H)-	30. 4 - (p - Fluorobenzyl) - 2 - [N-			
	phthalazinone hydrochloride hydrate.	methylpyrrolidinyl - (3)l - 1 (2H) - phthalazi-			
	F.p.: 275—278° C.	none.			
	21. 4 - (p - Aminobenzyl) - 2 - [N-	M.p.: 90—93° C.			
25	methyl - perhydroazepinyl - (4)l - 1 (2H)-	31. 4 - (p - Methylbenzyl) - 2 - [N-			
	phthalazinone dihydrochloride hydrate.	methylpyrolidinyl - (3)l - 1 (2H) - phthalazi-			
	F.p.: 270—277° C.				
	22. 4 - (p - Hydroxybenzyl) - 2 - [N-	none.			
	methyl perhydenareninyl (A) 1 (211)	M.p.: 96—98° C.			
30	methyl - perhydroazepinyl - (4) - 1 (2H)-	32. 4 - (p - Chlorobenzyl) - 2 - [perhydro-			
-	phthalazinone hydrochloride hydrate.	azepinyl - (4)l - 1 (2H) - phthalazinone			
	F.p.: 260—266° C.	fumarate.			
	Example 23.	M.p.: decomposition.			
	4 - (p - Chlorobenzyl) - 2 - [quinuclidinyl-	Example 33.			
35	(3)] - 1 (2H) - phthalazinone.	4 - (p - Chlorobenzyl) - 2 - [N-			
))	5.5 g. of α - (p - chlorophenyl) aceto-	methyl - perhydroazepinyl - (4)] - 1 (2H)-			
	phenone-o-carboxylic acid are dissolved in	phthalazinone.			
	30 cc. of 2N soda lye and 30 cc. of water.	1.0 g. of 4-(p-chlorobenzyl)-2-[perhydro-			
	4.3 g. of 3-quinuclidinyl-hydrazine dihydro-	azepinyl-(4)l-1 (2H)-phthalazinone are heated			
	chloride are added thereto and the mixture is	to boiling for 5 hours with 10 = -6 - 400/			
40	heated to boiling for 3 hours under an at-	to boiling for 5 hours with 10 g. of a 40%			
	mosphere of nitrogen. Upon cooling, a highly	aqueous formaldehyde solution and 11.6 g. of			
	viscous red oil is separated which crystallizes	formic acid. The solution is evaporated and			
	upon scratching. The solid metarial is flagged	the residue is triturated with dilute soda lye.			
	upon scratching. The solid material is filtered	The insoluble material is dissolved in chloro-			
45	off, washed with water and recrystallized.	form and the chloroform solution is dried and			
• • •	7.4 g. or 4-(p-chorobelizy)-2-[quiluclidiny]-	evaporated. The residue is dissolved in ether.			
	(3)]-1 (2H)-phthalazinone are obtained. This	0.8 g. of the hydrochloride is precipitated by			
	product melts at 181—182° C.	the addition of ethereal hydrochloric acid.			
		After recrystallization from alcohol, the com-			
	Example 24.	pound melts at 225—229° C.			
	4-(p-Chlorobenzyl)-2-[N-methylpiperidyl-	This compound is identical to the final			
50	(4)]-1 (2H)-phthalazinone.	product obtained according to Example 9.			
	11 g. of α -(p-chlorophenyl)acetophenone-o-	The following compound has been asset			
	carboxylic acid are dissolved in 120 cc. of	The following compound has been prepared			
	ethyl alcohol. A solution of 8 g. of N-	in a manner which is similar to the procedure			
	methylpiperidyl-(4)-hydrazine dihydrochloride	of Example 33.			
55	is added themse and the minutes is bessel as	34. 2 - [N - Methyl - perhydroazepinyl-			
	is according the matter is heated to	(4) - 4 - (p - trifluoromethylbenzyl) - 1			
	boiling for 8 hours under an atmosphere of	(2H)-phthalazinone.			
	nitrogen. The alcohol is distilled off and the	Example 35.			
	residue is triturated with dilute soda lye. The	4 - (p - Chlorobenzyl) - 2 - [N - methyl-			
	insoluble oily product is dissolved in chloro-	piperidyl - (3)l - 1 (2H) - phthalazinone.			
60	form and the chloroform solution is washed	4.9 g. of $3 - [4 - (p - \text{chlorobenzyl})] - \text{oxo-}$			
	and dried. Upon evaporation, 8.4 g. of the	phthalazinyl-(2)l-1-methyl-pyridinium iodide			
	phthalazinone base are obtained. The sumarate	are subjected to hydrogenation in 300 cc. of			
	melts at 191—193° C.	ethyl alcohol in the mesons of D.O.			
	The following compounds have been pre-	ethyl alcohol in the presence of PtO, as cat-			
	with the country make occur pic-	alyst for 7 hours at 80° C. and at a hydrogen			
		<u>.</u>			

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pressure of 100 atmospheres. The catalyst is filtered off and the alcohol is distilled off. The residue is treated with dilute soda lye and the insoluble materials are dissolved in methylene chloride. The methylene chloride solution is washed with water and dried over potash. The solvent is filtered off and the solid residue is recrystallized from 60 to 70% ethyl alcohol. The yield is 2.5 g.

10 M.p.: 154—156° C.

The following compounds have been prepared as described in Example 35:

36. 4 - (p - Methylbenzyl) - 2 - [N-methylpiperidyl - (3)] - 1 (2H) - phthalazinone.

M.p.: 137—139° C.

37. 4 - (p - Methoxybenzyl) - 2 - [N-methylpiperidyl - (3)l - 1 (2H) - phthalazinone.

20 M.p.: 87—93° C.

Example 38.

Tablets containing the products according to the present invention are prepared according to the following recipe as exemplified by the compound of Example 18:

	Active ingredient according		
30	to Example 19	1.0	mg.
	corn starch	51.0	
	secondary calcium phosphate,		
	anhydrous	20.0	mg.
	lactose	20.0	
	polyvinylpyrrolidone		mg.
	talcum		mg.
	magnesium stearate		mg.
35		100.0	mg.

The active compound is dissolved together with the polyvinyl-pyrrolidone in 5 times the amount of chloroform. A homogeneous mixture of calcium phosphate, lactose and 60% of the corn starch are mixed therewith and granulated. The dried granulate sieved to a maximal particle size of 0.75 mm. is mixed with the remaining amount of corn starch, talcum and magnesium stearate for half an hour and the mixture is pressed to tablets weighing 100 mg. each and having a diameter of 6 mm.

Example 39.

As described in Example 38, dragee-kernels weighing 100 mg., having a diameter of 6 mm. and a camber diameter of 5 mm., are prepared. These kernels are coated with a usual dragee coating in an amount of 170 mg.

Another batch of kernels is sprayed with a lacquer solution instead of the dragee coating. The resulting lacquer coating comprises:

55

Hydroxypropylmethyl		
cellulose	1.6 mg.	60
ethyl cellulose	0.5 mg.	•
polyethyleneglycol 4000	0.4 mg.	•
1,2-propylene glycol	0.25 mg.	
titanium dioxide	0.25 mg.	

The above recipes of the Examples 28 and 29 may be further followed by using a smaller amount of the active ingredient, such as 0.6 and 0.3 mg. instead of 1 mg. The difference in weight is balanced by additional amounts of corn starch.

Example 40.

1 g. of the hydrochloride of active ingredient of Example 9 are milled to a particle size of less than 75μ. The resulting product is mixed slowly with 999 g. of molten suppository fat at 40° C. with rapid stirring. The homogeneous mixture is poured into suppository molds to give suppositories each weighing 1.0 g. In an analogous manner, suppositories may be prepared containing 0.5 mg., 2 mg. or 6 mg. of active ingredient.

Exaxmple 41.

300 mg. of the hydrochloride corresponding to the product of Example 18 are dissolved together with 855 mg. of sodium chloride in 90 cc. of water and the solution is made up to 100 cc. The resulting solution is thoroughly filtered and poured into ampoules each measuring 1.1 cc. The closed ampoules are sterilized in an autoclave with steam under pressure at at least 120° C. for half an hour.

WHAT WE CLAIM IS: -

1. A basically substituted benzyl-phthalazinone derivative of formula I

$$(R_2)_n$$
 $(R_1)_m$
 $(R_2)_n$
 $(X)_p$
 $(X)_p$

wherein R₁ and R₂, which may be identical or different from each other, represent a hydrogen or halogen atom, an alkyl radical containing from 1 to 4 carbon atoms, an alkoxy radical containing from 1 to 4 carbon 100 atoms, or a hydroxy; trifluoromethyl, nitro or substituted or unsubstituted amino group, X is an alkylene group having the formula

m and n, which may be identical or different from each other, represent 1, 2 or 3, p is 0 or 1, and the group

-CH Y

is an N—C₁₋₄ alkyl-substituted pyrrolidinyl, an N—C₁₋₄-alkyl-substituted perhydroazepinyl, quinuclidinyl, tropanyl or scopyl group, the tropanyl or scopyl group being connected to the 2-nitrogen atom of the phthalazinone directly by way of a ring carbon atom of this tropanyl or scopyl group, and the other groups being connected to the 2-nitrogen atom of the phthalazinone directly or by way of an alkylene group of the formula

and their physiologically acceptable acid addition salts.

2. A basically substituted benzyl-phthalazinone derivative as claimed in claim 1 wherein R₁ and R₂ represent hydrogen, halogen, hydroxy, an alkyl group containing from 1 to 4 carbon atoms, an alkoxy group containing from 1 to 4 carbon atoms or trifluoromethyl, and m and n are 1 or 2.

3. A basically substituted benzyl-phthalazinone derivative as claimed in claim 1 or 2 wherein R₂ is hydrogen.

4. A basically substituted benzyl-phthalazinone derivative as claimed in any of claims 1 to 3 wherein R₁ is a hydrogen, fluorine, chlorine or bromine atom or a methoxy, ethoxy, methyl, hydroxy or trifluoromethyl group, R₂ is a hydrogen atom, m is 1 or 2, p is 0 and the group

is the N-methyl-perhydroazepinyl, the tropanyl or the quinuclidinyl group.

5. A basically substituted benzyl-phthalazinone derivative as claimed in claim 4 wherein R₁ is p-chloro or p-fluoro and the group

is the N-methyl-perhydroazepinyl-(4) group.
6. A process for the production of a basically substituted benzyl-phthalazinone derivative as

claimed in any of claims 1 to 5, comprising A) subjecting a compound of Formula II

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or a reactive derivative thereof, wherein R₁, R₂, m and n have the same meanings as in claims 1 to 5, to reaction with a hydrazine compound of formula III

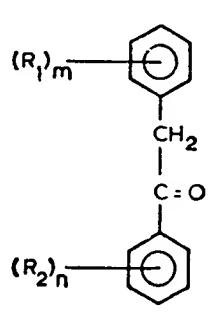
III

wherein R₃ is hydrogen or the group

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X, p and

having the same meanings as in claims 1 to 5, or 60 B) subjecting a compound of formula IV



IV

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wherein R₁, R₂, m and n have the same meanings as in claims 1 to 5 to reaction with a compound of formula V

 $J_{\mathbf{r}}$

V

wherein R₃ has the same meaning as in formula III and R₄ is an alkyl group containing from 1 to 4 carbon atoms,

C) subjecting a compound of formula VI

VI

wherein R₂ and n have the same meanings as in claims 1 to 5 and R₃ has the same meaning as in formula III, to reaction with a compound of formula VII

VII

wherein R₁ and m have the same meanings as in claims 1 to 5, and Z is a halogen atom or a hydroxy or alkoxy group, or
D) when



is an N—C₁₋₄ alkyl-substituted pyrrolidinyl,
N—C₁₋₄ alkyl-substituted piperidyl or
N—C₁₋₄ alkyl-substituted perhydroazepinyl
group,
subjecting a compound of formula I as defined
in claim 1, wherein the group

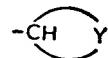


is a pyrrolidinyl group, a piperidyl group or a perhydroazepinyl group to reaction with an alkylating agent, containing 1 to 4 carbon atoms, and subjecting a benzyl-phthalazinone derivative of formula XIII

wherein R₁, R₂, m and n have the same meaning as in formula I and R₃ is hydrogen, resulting from process A), B) or C), to reaction with a compound of formula VIII

wherein Q represents an atom or group which, upon substitution of the amide group, is split off together with its electron doublet and R₃ is the group

X, p and



having the same meanings as in claims 1 to 5, and converting the thus obtained benzylphthalazinone derivative, if desired, with an appropriate acid into a physiologically acceptable acid addition salt thereof or, if desired, converting a resulting salt of a benzylphthalazinone derivative into the free base.

7. A process as claimed in claim 6 wherein Q is a halogen atomor a sulphonic ester group.

8. The final product of each of the individual Examples 1—37 herein.

9. A process as claimed in claim 6 substantially as described with reference to any of Examples 1 to 37.

ELKINGTON AND FIFE,
Chartered Patent Agents,
High Holborn House,
52/54 High Holborn, London, WC1V 6SH,
Agents for the Applicants.

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